**O.P.Code:** 23HS0830

**R23** 

H.T.No.

#### SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 LINEAR ALGEBRA & CALCULUS

(Common to all Branches)

Time: 3 Hours

PART-A

(Answer all the Questions  $10 \times 2 = 20 \text{ Marks}$ )

1 a

Find the rank of the matrix  $A = \begin{bmatrix} 1 & -1 & 2 & -1 \\ 3 & 1 & 0 & 1 \end{bmatrix}$ .

**b** Determine k such that the system of homogeneous equations 2x + y + z = 0, x + y + 3z = 0, 4x + 3y + kz = 0 has non-trivial solution.

c Let  $\lambda$  be an Eigen value of A and X be its corresponding eigenvector.

Then, Sow that  $A^{-1}$  (if it exists) has Eigen value  $\frac{1}{2}$ .

d Convert the symmetric matrix  $\begin{bmatrix} 1 & 2 & 3 \\ 2 & 1 & 3 \end{bmatrix}$  into the quadratic form.

e State Lagrange's mean value theorem.

Find c of Lagrange's mean value theorem for  $f(x) = x^2 \ln[1, 5]$ .

Evaluate  $\lim_{\substack{x \to 1 \ y \to 2}} \frac{2x^2y}{x^2+y^2+1}$ 

h Define Extreme value of a function of two variables.

If  $x = r \cos \theta$ ,  $y = r \sin \theta$ , find  $\frac{\partial(x, y)}{\partial(r, \theta)}$ .

**j** Evaluate  $\int_0^2 \int_0^x y \, dy \, dx$ 

**PART-B** 

(Answer all Five Units  $5 \times 10 = 50$  Marks)

UNIT-I

Solve the following system of equations by Gauss-Jacobi Iteration method 2

27x + 6y - z = 85; x + y + 54z = 110; 6x + 15y + 2z = 72.

Reduce the matrix  $A = \begin{bmatrix} 1 & 2 & 3 & 2 \\ 2 & 3 & 5 & 1 \\ 1 & 3 & 4 & 5 \end{bmatrix}$  into Echelon form and find its 3

rank?

**b** Solve completely the system of equations

4x + 2y + z + 3w = 0; 6x + 3y + 4z + 7w = 0; 2x + y + w = 0.

Verify Cayley Hamilton theorem for  $A = \begin{bmatrix} 1 \\ 2 \\ 2 \end{bmatrix}$ -2 and find

 $A^{-1}$  and  $A^{4}$  using Cayley Hamiltion theorem

OR

Max. Marks: 70

CO<sub>1</sub> L<sub>2</sub> 2M

CO<sub>2</sub> L2 2M

CO<sub>2</sub> L2 2M

CO<sub>3</sub> L12M

CO<sub>3</sub> **L2** 2M

**CO4 L5** 2M

L1 CO<sub>5</sub> 2ML<sub>2</sub> **CO6** 2M

**CO6 L5** 2M

CO<sub>2</sub> L3 10M

CO<sub>1</sub> L3

CO<sub>1</sub>

**5M** 

\*\*\* END \*\*\*

O.P.Code: 23ME0302 R23 H.T.No.

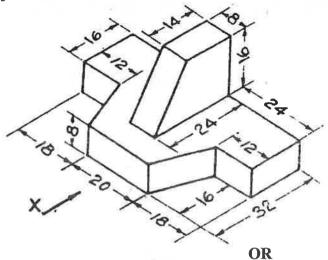
### SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 ENGINEERING GRAPHICS

ENGINEERING GRAPHICS						
Tim	(Common to CSE, ECE, EEE, CSIT) ne: 3 Hours	May	Mar1	ks: 70		
	(Answer all the Questions $5 \times 14 = 70$ Marks)	wax.	wall	25. 70		
4	UNIT-I					
1	Construct an ellipse, with distance of the focus from the directrix as 50 mm	CO <sub>1</sub>	<b>L6</b>	14M		
	and eccentricity as 2/3. Also draw normal and tangent to the curve at a point					
	40 mm from the directrix.					
	OR					
2	Construct a cycloid, given the diameter of the generating circle as	CO <sub>1</sub>	<b>L6</b>	14M		
F.	40mm.Draw a tangent and normal to the curve at a point on it,35mm from					
	the base line.					
	UNIT-II					
3	Draw the projections of the following points, keeping the distance between	CO <sub>2</sub>	L1	14M		
_	the projectors as 25mm on the same reference lines.	CO2		14141		
	A – 20mm above HP and 30mm in front of VP					
	B – 20mm above HP and 30mm behind VP					
	C – 20mm below HP and 30mm behind VP					
	D – 20mm below HP and 30mm in front of VP					
	E – On HP and 30mm in front of VP					
	F – On VP and 20mm above HP					
	G – Lying on both HP and VP					
	OR					
4	A line AB of 100mm length is inclined at an angle of 300 to HP and 450 to	CO2	Τ 1	1.4N/I		
7	VP. The point A is 15mm above HP and 20mm in front of VP. Draw the	COZ	LI	14M		
	projections of the line					
_	UNIT-III					
5	a Draw the projections of a cylinder of base 30mm diameter and axis	CO <sub>3</sub>	L6	<b>7M</b>		
	50mm long, when it is resting on H.P on one of its bases.					
	<b>b</b> Draw the projections of a cone of base 30mm diameter and axis 50mm	CO <sub>3</sub>	L6	<b>7M</b>		
	long, when it is resting on H.P on one of its bases.					
	OR					
6	A cone of diameter 50 mm and axis 60 mm has its generator in the VP and	CO <sub>3</sub>	<b>L6</b>	14M		
	the axis is parallel to the HP. Draw its projections.					
	UNIT-IV					
7	A pentagonal pyramid with edge of base 25 mm and axis 65 mm long, its	CO <sub>4</sub>	<b>L6</b>	14 <b>M</b>		
	base is resting on HP. It is cut by a section plane, inclined at 600 to HP and					
	perpendicular to VP it bisects the axis. Draw the projections and obtain the					
	true shape of the section.					
	OR					
8	A cone of base 50 mm diameter and height 65 mm rests with its base on HP.	CO <sub>4</sub>	L1	14M		
	A section plane perpendicular to VP and inclined at 300 to HP bisects the	_ •				
	axis of the cone. Draw the development of the lateral surface of the					
	truncated cone.					

UNIT-V

9 Draw three views of the blocks shown pictorially in figure according to first CO6 L6 14M angle projection



Draw the isometric view of a hexagonal prism of base side 30 mm and axis CO5 L1 14M 70mm. The prism rests on its base on the HP with an edge of the base parallel to the VP.

\*\*\* END \*\*\*



**O.P.Code:** 23CE0101

**R23** 

H.T.No.

### SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 BASIC CIVIL & MECHANICAL ENGINEERING

(Common to CE, ME, CSM, CIC, CAD, CCC & CAI)

\*Note: Answer PART-A from pages 2 to 20 and PART-B from 21 to 39.

Time: 3 Hours		Max.	Mark	ks: 70
	<u>PART-A</u> (CIVIL)			
	(Answer all the Questions $5 \times 1 = 5$ Marks)			
1	a List various sources of water.	CO <sub>1</sub>	L2	<b>1M</b>
	<b>b</b> What is means by surveying.	CO <sub>2</sub>	L1	1 <b>M</b>
	<b>c</b> What are the uses of contour mapping.	CO <sub>2</sub>	L1	1 <b>M</b>
	d Differentiate between true meridian and magnetic meridian.	CO <sub>3</sub>	L1	1M
	e What are the reasons to build a tunnel?	CO <sub>3</sub>	L1	1 <b>M</b>
	(Answer all Three Units $3 \times 10 = 30$ Marks) (CIVIL)			
	UNIT-I			
2	a List out various forms of steel used in construction. Explain briefly.	CO <sub>1</sub>	<b>L2</b>	<b>5M</b>
	<b>b</b> Which are steel channel sections available in the market? Give neat	CO <sub>1</sub>	L1	<b>5M</b>
	sketches.			
	OR			
3	a Write about good qualities of cement.	CO <sub>1</sub>	L1	5M
	<b>b</b> List out grades of cement and their uses.	CO <sub>1</sub>	L1	<b>5M</b>
	UNIT-II			
4	The following staff readings were observed successively with a level, the	CO <sub>2</sub>	L3	10M
	instrument having been moved after third, sixth and eighth readings:			
	2.230; 1.610; 0.980; 2.090; 2.865; 1.265; 0.600; 1.980; 1.045; 2.685			
	metres.Enter the above readings in a page of a level book and calculate the			
	R.L. of points by rise and fall method. The R.L. of B.M. is 100.000m.			
	OR			
5	Explain briefly the various types of levelling instruments.	CO <sub>2</sub>	<b>L2</b>	<b>10M</b>
	UNIT-III			
6	Briefly discuss about quality of water. What are the important requirements	CO <sub>3</sub>	<b>L2</b>	10M
	of water for domestic use?			
	OR			
7	a What are the purposes for constructing a dam?	CO <sub>3</sub>	L1	5 <b>M</b>
	<b>b</b> Explain briefly about how dams are classified according to material use.	CO <sub>3</sub>	L2	5M

 $\frac{PART-B}{(MECHANICAL)}$ (Answer all the Questions  $5 \times 1 = 5$  Marks)

		( The state of the			
1	f	Define Strength of a material.	CO <sub>1</sub>	L1	<b>1M</b>
	g	Explain the Role of Ceramics in Engineering?	CO <sub>1</sub>	L1	<b>1M</b>
	h	What are smart materials and mention examples.	CO <sub>2</sub>	L1	<b>1M</b>
	i	Define CNC Machining.	CO <sub>2</sub>	L1	<b>1M</b>
	j	List out the basic components of Robot.	CO <sub>3</sub>	L1	1 <b>M</b>
		(Answer all Three Units $3 \times 10 = 30$ Marks) (MECHANIC	CAL)		
		UNIT-IV			
8		Explain the Role of mechanical engineering in society?	CO <sub>1</sub>	L1	<b>10M</b>
		OR			
9	a	Draw the flow chart classifying engineering materials.	CO <sub>1</sub>	L1	5M
	b	Differentiate between metals and Nonmetals.	CO <sub>1</sub>	L1	5M
		UNIT-V			
10	a	Illustrate the working of Four stroke diesel engine with a neat sketch	CO <sub>2</sub>	<b>L2</b>	<b>5M</b>
	b	Differentiate between two stroke engine and four stroke engine	CO <sub>2</sub>	<b>L2</b>	<b>5M</b>
		OR			
11		What is 3D printing and explain the advantages of 3D printing?	CO <sub>2</sub>	L1	<b>10M</b>
9		UNIT-VI			
<b>12</b>		Illustrate the working of steam power plant with a neat sketch.	CO <sub>3</sub>	<b>L2</b>	<b>10M</b>
		OR			
13	a	Explain applications of robots in various industries.	CO <sub>3</sub>	<b>L2</b>	<b>5M</b>
	b	Explain the main Robot anatomy with neat sketch.	CO <sub>3</sub>	<b>L2</b>	<b>5M</b>

O.P.Code:23CS0501

R23

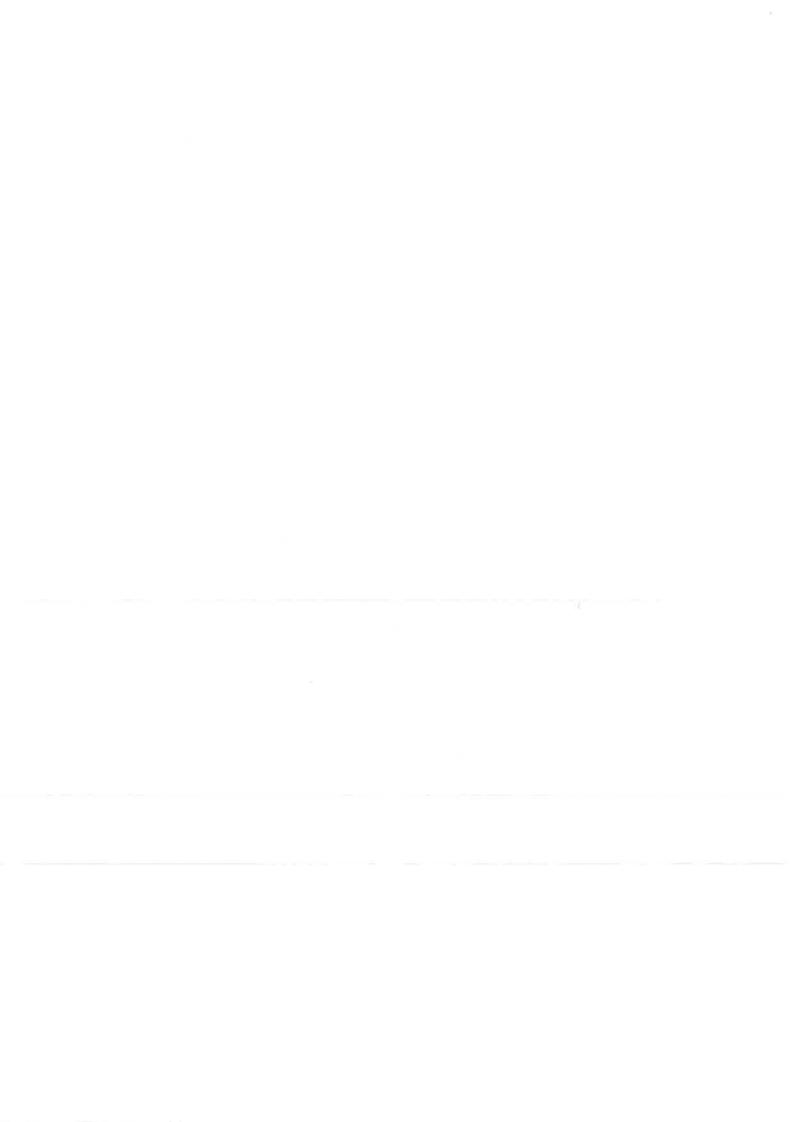
H.T.No.

# SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 INTRODUCTION TO PROGRAMMING

(Common to All)

Time: 3 Hours (Common to All)		Max.	Mark	s: 70	
	PART-A				
		(Answer all the Questions $10 \times 2 = 20 \text{ Marks}$ )			
1	a	List the different flow chart symbols.	CO1	L1	<b>2M</b>
	b	What is meant by type conversion?	CO1	L1	2M
	c	Summarize break and continue keyword.	CO <sub>2</sub>	<b>L2</b>	2M
	d	Write the syntax for nested if else statement.	CO <sub>2</sub>	L1	<b>2M</b>
	e	What is a String? Give example.	CO <sub>3</sub>	L1	<b>2M</b>
	f	List the different string handling functions	CO <sub>3</sub>	L1	<b>2M</b>
	g	Compare Arrays and Pointers.	CO4	<b>L4</b>	<b>2M</b>
	h	What is meant by structure and write the syntax for structure declaration.	CO6	L1	<b>2M</b>
	i	What is meant by call-by-value?	CO <sub>5</sub>	L1	<b>2M</b>
	j	List the different file operations in C.	CO <sub>6</sub>	L1	2M
		<u>PART-B</u>			
		(Answer all Five Units $5 \times 10 = 50$ Marks)			
		UNIT-I			
2	a	Define a flow chart. List the different symbols in flowchart.	CO <sub>1</sub>	L1	5M
	b	Draw a flowchart for computing the Simple interest.	CO <sub>1</sub>	<b>L2</b>	<b>5M</b>
		OR			
3	a	What is meant by data type. List the different data types with their sizes.	CO <sub>1</sub>	L1	<b>5M</b>
	b	Define a variable. Write the variable declaration. What are the rules for	CO <sub>1</sub>	L1	5M
		declaring a variable?			
		UNIT-II			
4	a	Write the syntax and illustrate the following statements with example	CO2	L3	5M
		i)if Statement ii) if else Statement iii) else if ladder iv) Nested if			
		statements v) Switch Case.			
	b	Develop a 'C' program to perform the basic arithmetic operations using	CO <sub>2</sub>	<b>L6</b>	5M
		switch case statement.			
		OR			
5	a	Explain the For Loop with syntax and example.	CO <sub>2</sub>	L2	5M
	b	Compose a C program to print following series	CO <sub>2</sub>	<b>L6</b>	5M
		1			
		2 2			
		3 3 3			
	-	4 4 4 4			
		UNIT-III			
6	a	Explain the One-Dimensional array with example.	CO2	L2	5M
	b	Compose a C program for Transpose of a given matrix.	CO2	L6	5M
		OR			



O.P.Code: 23HS0803

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H.T.No.

### SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 ENGINEERING CHEMISTRY

(Common to CE & ME)

	(Common to CE & ME)			
Tim	e: 3 Hours	Max.	Mark	ks: 70
	PART-A			
1	(Answer all the Questions 10×2=20 Marks)	004		
1	a What are scales and sludge's?	CO1	L1	2M
	b List two disadvantages of using hard water in boilers.	CO1	L1	<b>2M</b>
	c Differentiate electrochemical cell and electrolytic cell.	CO <sub>2</sub>	<b>L2</b>	<b>2M</b>
	d What is Pilling Bedworth rule?	CO <sub>2</sub>	L1	<b>2M</b>
	e What are additional polymers? Give one example.	CO <sub>3</sub>	L1	<b>2M</b>
	f What is meant by combustion of fuels?	CO4	L1	<b>2M</b>
	g What are composite materials?	CO <sub>5</sub>	L1	<b>2M</b>
	h Define cloud point.	CO <sub>5</sub>	L1	<b>2M</b>
	i What is Freundlich's adsorption isotherm?	CO <sub>6</sub>	L1	<b>2M</b>
	j What is physical adsorption?	CO <sub>6</sub>	L1	<b>2M</b>
	PART-B			
	(Answer all Five Units 5×10=50 Marks)			
	UNIT-I			
2	a Describe the ion-exchange process for demineralization of water.	CO <sub>1</sub>	<b>L2</b>	<b>6M</b>
	<b>b</b> What are the advantages and disadvantages of ion-exchange process?	CO <sub>1</sub>	L1	<b>4M</b>
	OR			
3	a Discuss the causes and prevention of priming and foaming in boiler	CO <sub>1</sub>	<b>L2</b>	5M
	b Explain in detail about the specifications of drinking water.	CO <sub>1</sub>	<b>L2</b>	5M
	UNIT-II			
4	a Describe the construction and working of Hydrogen-Oxygen fuel cell	CO2	L2	5M
	b Explain the charging and discharging reactions of Lithium-ion battery.	CO <sub>2</sub>	L2	5M
	OR	002	~-	0111
5	a What is cathodic protection? Explain the sacrificial anodic protection.	CO <sub>2</sub>	L2	5M
	b Describe the mechanism of differential aeration corrosion with example.	CO <sub>2</sub>	L2	5M
	UNIT-III	002		5111
6	a What is functionality of a monomer? Explain its significance.	CO3	L2	5M
U	b Discuss about the synthesis and properties of Polystyrene.	CO3	L2 L2	5M
	OR	COS	LL	21/1
7	a Discuss the ultimate analysis of coal with its significance.	CO4	L2	5M
í	b Explain the importance of blended petrol using ethanol as fuel.	CO4	L2	5M
	UNIT-IV	CO4		3111
8		COF	т 2	10%/
o	Classify and discuss the functions and properties of lubricating oils with suitable examples.	CO <sub>5</sub>	L2	10M
	OR			
9	a Explain the factors affecting the refractory materials.	CO5	L2	5M
1	b Explain about setting and hardening of Portland cement.	CO5	L2	5M
1		COS	L4	3111
10	Explain the preparation of none metals by chamical and alectural ancient	001	Τ.Δ	107 #
10	Explain the preparation of nano-metals by chemical and electrochemical	CO <sub>6</sub>	L2	<b>10M</b>
	method.			
11	OR  a Discuss I anomair adsoration isotherm	COC	т о	<i>E</i> 1
11	a Discuss Langmuir adsorption isotherm.	CO6	L2	5M
	<b>b</b> Explain about the stabilization of colloids by Solid-Liquid Interface.	CO <sub>6</sub>	<b>L2</b>	5M
	*** END ***			

Q.P.Code: 23HS0801

R23

H.T.No.

## SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 **CHEMISTRY**

(Common to CSM, CIC, CAD, CCC & CAI)

		(Common to CSM, CIC, CAD, CCC & CAI)			
Tim	e: :	B Hours	Max.	Mark	s: 70
		<u>PART-A</u>			
		(Answer all the Questions $10 \times 2 = 20 \text{ Marks}$ )			
1	a	State Heisenberg Uncertainty principle and mention the terms.	<b>CO1</b>	L2	2M
	b	Find the bond order of CO molecule based on MOT.	CO <sub>1</sub>	L3	2M
	c	Define Semi Conductor. Give any two examples.	CO <sub>2</sub>	<b>L2</b>	2M
	d	What are Nano particles? Mention its applications.	CO <sub>2</sub>	L1	2M
	e	Define Electro Chemical Sensor and write its Classification.	CO <sub>3</sub>	L2	<b>2M</b>
	f	Define Fuel cells. Name any two fuel cells.	CO <sub>3</sub>	<b>L2</b>	2M
	g	Identify Monomers present in Bakelite and Nylon-6,6	CO4	L2	2M
	_	List out the applications of Conducting Polymers.	CO4	L1	2M
	i	Give the selection rules associated with IR Spectroscopy.	CO5	L2	2M
	j	Define Stationary phase and Mobile Phase.	CO5	L1	2M
	J	PART-B	COS		2111
		(Answer all Five Units $5 \times 10 = 50$ Marks)			
		UNIT-I			
2		Derive Schrodinger Wave equation.	CO <sub>1</sub>	L3	<b>5M</b>
	b	Give the important postulates of Molecular orbital theory	CO <sub>1</sub>	L2	<b>5M</b>
		OR			
3	a	Illustrate the energy diagram and bond order of CO molecule.	CO <sub>1</sub>	L2	5M
	b	Explain $\pi$ - molecular orbital of 1, 3- Butadiene with a neat sketch.	CO <sub>1</sub>	L2	5M
		UNIT-II			
4	а	Explain in detail about classification and doping of semiconducting	CO2	<b>L2</b>	5M
-		materials.	CO2		SIVI
	b	Explain the Classification of Super capacitors with suitable examples.	CO2	L2	5M
	I.	OR	COZ	LL	SIVI
5			CO1	т 2	
3		Discuss the properties and applications of Nano materials.	CO2	L2	5M
	D	Describe the classification and properties Graphine nanoparticles.	CO <sub>2</sub>	L2	5M
		UNIT-III			
6	a	Discuss the titration curves obtained in weak acid and strong base with	CO <sub>3</sub>	L2	5M
		suitable example.			
	b	Explain the reactions involved in Li-Ion Secondary Batteries.	CO <sub>3</sub>	<b>L2</b>	5M
		OR			
7	a	With a neat sketch explain Hydrogen-Oxygen fuel cell.	CO3	L3	6 <b>M</b>
	b	Explain about potentiometric redox titrations with example.	CO <sub>3</sub>	<b>L2</b>	<b>4M</b>
		UNIT-IV			
8	я	Describe the Anionic and Cationic Polymerization mechanism.	CO4	<b>L2</b>	6M
Ū		Explain synthesis and applications of Poly tetra fluoro ethylene(PTFE).	CO4	L2	4M
	U	OR	CO4	LZ	4111
9			COA	т о	53.E
7		Describe the preparation, properties and applications of Buna-N & S.	CO4	L2	5M
	D	Give a detailed note on Bio – degradable polymers.	CO4	L2	5M
		UNIT-V			
10	a	Explain the electronic transitions of UV-Visible Spectroscopy.	CO <sub>5</sub>	<b>L2</b>	<b>5M</b>
	b	Explain in detail different regions and Instrumentation IR Spectroscopy.	CO <sub>5</sub>	<b>L2</b>	<b>5M</b>
		OR			
11	a	Discuss the principle and instrumentation of HPLC chromatography.	CO <sub>5</sub>	L2	<b>5M</b>
		List out the important applications of HPLC.	CO <sub>5</sub>	L1	5M
		*** END ***			

O.P.Code: 23HS0840

classical free electron theory.

**R23** 

H.T.No.

### SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

#### B. Tech I Year I Semester Supplementary Examinations July/August-2024 **ENGINEERING PHYSICS**

(Common to CSE, ECE, EEE & CSIT) Time: 3 Hours Max. Marks: 70 PART-A (Answer all the Questions  $10 \times 2 = 20$  Marks) 1 a Define interference. CO<sub>1</sub> L1 2MWhat is plane polarized light? **CO1** L<sub>1</sub> 2M**c** What are lattice parameters? CO<sub>2</sub> L<sub>1</sub> 2Md Draw the planes for given Miller indices (111) and (202) in cubic system. CO<sub>2</sub> L3 2Me Define dielectric polarization. **CO3** L<sub>1</sub> 2Mf What is Hysteresis? CO<sub>4</sub> L<sub>1</sub> 2M**g** What are matter waves? **CO5** L1 2M**h** What is Fermi energy level? CO<sub>5</sub> L1 2MWhat is extrinsic semiconductor? **CO6** L1 **2M** What are the drift and diffusion current densities in semiconductors? **CO6** L1 2MPART-B (Answer all Five Units  $5 \times 10 = 50$  Marks) UNIT-I a State and explain principle of superposition. 2 **CO1** L2 **4M b** Discuss the theory of interference of light due to thin films by reflection CO<sub>1</sub> **6M** with suitable ray diagram. OR 3 a Explain the Polarization by reflection with neat sketch. CO<sub>1</sub> L<sub>2</sub> **5M b** Illustrate the Double refraction in crystal. CO<sub>1</sub> **L4 5M** UNIT-II a What are Miller indices? Mention the procedure to find Miller indices CO<sub>2</sub> L1 **6M b** Write the important features of Miller indices CO<sub>2</sub> L1 **4M** a Explain how crystal structure determined by Laue X-Ray diffraction 5 CO<sub>2</sub> L2 **7M** method. **b** What are the advantages of Laue X-Ray diffraction method? CO<sub>2</sub> L<sub>1</sub> **3M** UNIT-III a Obtain Clausius-Mosotti equation and explain how it can be used to CO<sub>3</sub> **L4 7M** determine the dipole moment of a polar molecule. **b** A solid elemental dielectric with 3x1028 atoms/ m3 shows an electronic CO3 L1 3M polarisability of 10-40 F-m 2 assuming the internal electric field to be a Lorenz field. Calculate a dielectric constant of the material. a Explain hysteresis of ferromagnetic material. CO<sub>4</sub> L2 **5M b** Distinguish between soft and hard magnetic materials. CO<sub>4</sub> L<sub>2</sub> 5M UNIT-IV a Derive Schrödinger's time independent wave equation CO<sub>5</sub> L3 **7M b** Explain the physical significance of wave function **CO5 L2** 3Ma What are the postulates of classical free electron theory? CO<sub>5</sub> L1 **4M** b Derive an expression for electrical conductivity in a metal by using

**6M** 

CO<sub>5</sub>

**L3** 

			-
ш	N	ш.	-V
	T 4.1		- ▼

<b>10</b>	a	Derive the expression for the conductivity of intrinsic semiconductor.	<b>CO6</b>	L3	7M
	b	The following data are given for an intrinsic Ge at 300K. Calculate the conductivity of the sample? ( $n_i$ = 2.4 x10 $^{19}$ m $^{-3}$ , $\mu_e$ = 0.39 m $^2$ -V $^{-1}$ S $^{-1}$ , $\mu_p$ = 0.19 m $^2$ -V $^{-1}$ S $^{-1}$ )	CO6	L3	3M
		OR			
11	a	Describe the Hall Effect in semiconductors.	CO <sub>6</sub>	L2	8M
	b	What are the applications of Hall Effect?	<b>CO6</b>	L1	2M
		*** END ***			



O.P.Code: 23EE0201

Time: 3 Hours

**R23** 

H.T.No.

#### SIDDHARTH INSTITUTE OF ENGINEERING & TECHNOLOGY:: PUTTUR (AUTONOMOUS)

### B.Tech I Year I Semester Supplementary Examinations July/August-2024 BASIC ELECTRICAL & ELECTRONICS ENGINEERING

(Common to CSE, ECE, EEE & CSIT)

\*Note: Answer PART-A from pages 2 to 20 and PART-B from 21 to 39.

**PART-A** (ELECTRICAL)

Max. Marks: 70

L1

1M

**1M** 

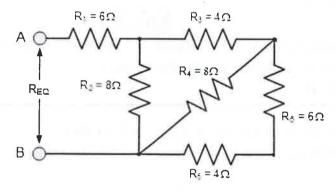
(Answer all the Questions  $5 \times 1 = 5$  Marks)

- 1 a Define Impedance. CO<sub>1</sub>
  - **b** List any Five parts of a Transformer. CO<sub>2</sub> L1 **1M c** What are the different types of Earthing?
  - CO<sub>2</sub> L1**1M** d Define Faradays law. CO<sub>3</sub> L1 **1M**
  - e What is a step-down transformer? CO<sub>3</sub> L1

(Answer all Three Units  $3 \times 10 = 30 \text{ Marks}$ ) (ELECTRICAL)

UNIT-I

- 2 a Find equivalent resistance when three resisters are connected in parallel. CO<sub>1</sub> L1 **4M** 
  - **b** Find the equivalent resistance for the circuit shown below. CO<sub>1</sub> L1 **6M**



OR

- 3 a Explain the concept of impedance in an A.C circuits. **CO1** L22M**b** Define the following CO<sub>1</sub> **L1 8M** 
  - i)Waveform, ii) Time period, iii) frequency, iv) Amplitude

UNIT-II

Draw and Explain the constructional diagram of a single phase 10M transformer in detail.

OR

- a Explain the operating principles of Moving Iron instruments CO<sub>2</sub> L2 **5M** CO<sub>2</sub> L<sub>2</sub>
  - b Determine the unknown resistance using Wheatstone bridge

**5M** 

UNIT-III a Define Earthing and explain the types of earthing

CO<sub>3</sub> **L5 6M** 

**b** What are the advantages of earthing?

CO<sub>3</sub> L1 **4M** 

7 What is solar power plant? Explain the operation with layout CO<sub>3</sub> L1 10M

### PART-B(ELECTRONICS)

(Answer all the Questions  $5 \times 1 = 5$  Marks)

		(Answer all the Questions $3 \times 1 - 3$ Marks)			
1	f	Define biasing.	CO <sub>4</sub>	L2	1M
	g	How PN diode is formed?	CO <sub>4</sub>	L1	1 <b>M</b>
	h	What is an emitter?	CO <sub>5</sub>	L1	<b>1M</b>
	i	List the names of universal gates with symbols	CO <sub>6</sub>	L1	<b>1M</b>
	j	What are the basic properties of Boolean algebra?	CO <sub>6</sub>	L1	<b>1M</b>
	•	(Answer all Three Units $3 \times 10 = 30$ Marks) (ELECTRONIC	CS)		
		UNIT-IV			
8	a	Define Zener diode and its characteristics.	CO <sub>4</sub>	L1	5M
	b	What is Zener effect?	CO <sub>4</sub>	<b>L2</b>	5M
		OR			
9		With the neat sketch, Explain the operation of an NPN transistor and PNP	CO <sub>4</sub>	<b>L5</b>	<b>10M</b>
		transistor.			
		UNIT-V			
10		Draw the block diagram of Electronic Instrumentation System and	CO <sub>5</sub>	<b>L2</b>	10M
		explain the function of each block.			
		OR			
11		Briefly explain the Working of Common Emitter Amplifier with proper	CO <sub>5</sub>	L5	<b>10M</b>
		circuit and wave forms.			
		UNIT-VI			
12	a	Explain differences between combinational and sequential circuits.	CO <sub>6</sub>	L2	<b>5M</b>
	b	Perform the following addition using excess-3 code	CO6	L3	<b>5M</b>
		i)386+756 ii)12+38			
		OR			
13	a	What is BCD codes and what are the various BCD codes	<b>CO6</b>	L1	<b>6M</b>
	b	Perform the following Decimal addition to 8421 BCD code.	<b>CO6</b>	<b>L4</b>	<b>4M</b>
		i)48+58, ii)186+237			

\*\*\* END \*\*\*